

NHC-Catalyzed Enantioselective [2 + 2] and [2 + 2 + 2] Cycloadditions of Ketenes with Isothiocyanates

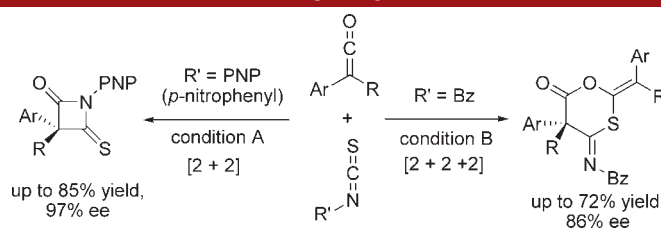
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ABSTRACT



The enantioselective *N*-heterocyclic carbene-catalyzed formal [2 + 2] and [2 + 2 + 2] cycloaddition of ketenes and isothiocyanates were developed. Reaction with *N*-aryl isothiocyanates at room temperature favors the [2 + 2] cycloaddition, while reaction with *N*-benzoyl isothiocyanates at $-40\text{ }^{\circ}\text{C}$ favors the [2 + 2 + 2] cycloaddition.

The cycloaddition reactions of ketenes are powerful tools for the construction of various cyclic compounds.^{1,2} During the past few years, we demonstrated that *N*-heterocyclic carbenes (NHCs)³ are efficient catalysts for cycloaddition reactions of ketenes with carbonyl compounds, imines, heterodienes, oxaziridines, carbon disulfide, and

N-sulfinylanilines.^{4,5} Because of their easy availability and diverse reactions, isothiocyanates are widely utilized, especially in the synthesis of heterocycles.^{6,7} In 1965, Winberg et al. reported the reaction of NHCs with isothiocyanates to form stable adducts.⁸ Recently, Cheng et al. have successfully developed this process for the synthesis of a

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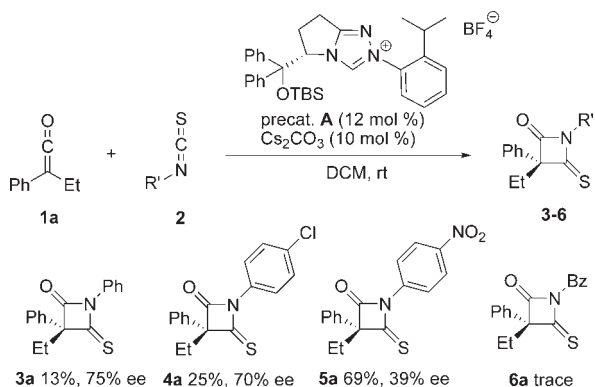
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wide range of heterocycles.⁹ Very recently, Yadav et al. reported the NHC-catalyzed [2 + 2] cycloaddition of isothiocyanates and nitroolefins.^{10,11} In this paper, we report an NHC-catalyzed enantioselective reaction of ketenes with isothiocyanates for the synthesis of 4-thioxo-2-azetidiones (thioxo- β -lactams), which are potential synthons for bioactive heterocycles,¹² such as sulfur-containing β -lactam antibiotics.¹³ Although 4-thioxo-2-azetidiones have been prepared from degradation of penicillin since 1976,¹⁴ to the best of our knowledge, the enantioselective synthesis of this key motif remains unrealized.

Initially, we found that the reaction of ketene **1a** and *N*-phenyl isothiocyanate gave the desired 4-thioxo-2-azetidione **3a** in 13% yield with 75% ee in the presence of NHC **A1'**, generated freshly from NHC precursor **A1** and Cs₂CO₃, at room temperature (Scheme 1). Encouraged by this result, several isothiocyanates were then screened. Reaction with *N*-*p*-chlorophenylisothiocyanates gave the desired product **4a** in 25% yield with 70% ee. The yield was increased to 69% albeit with 39% ee when *N*-*p*-nitrophenyl isothiocyanate was used. Reaction of *N*-benzoyl isothiocyanate gave no desired cycloadduct but a complex mixture.

Scheme 1. Screening of Isothiocyanates



The reaction of ketene **1a** and *N*-*p*-nitrophenyl isothiocyanate (**2c**) was optimized (Table 1). It is interesting that

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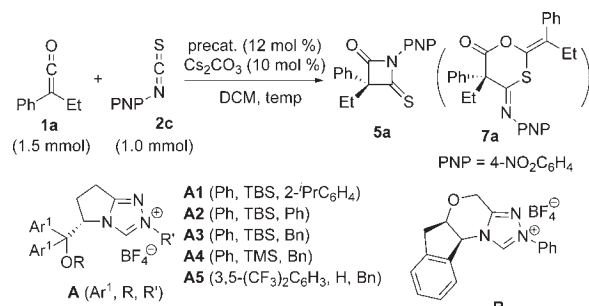
(12) Bachi, M. D.; Goldberg, O.; Gross, A.; Vaya, J. *J. Org. Chem.* **1980**, *45*, 1481–1485.

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better yield and enantioselectivity were realized when the reaction was carried out in dilute solution (entries 1–5). Several NHC precursors **A2–A5**, derived from *L*-pyroglutamic acid were then tested for the reaction. NHC precursor **A2** with an *N*-phenyl group led to a decreased yield with moderate enantioselectivity (entry 6), while NHC precursors **A3–A4** with an *N*-benzyl group showed better enantioselectivity but decreased yield (entries 7 and 8). NHC precursor **A5** with a free hydroxyl group afforded the cycloadduct **5a** in 73% with 96% ee (entry 9). NHC precursor **B**, derived from chiral indanol, resulted in low yield and enantioselectivity (entry 10). The yield was improved when excess ketene (2.5 equiv) was used and added via a syringe pump (entries 11 and 12). Unexpectedly, a [2 + 2 + 2] cycloadduct **7a** was observed, along with the normal [2 + 2] cycloadduct **5a**, when the reaction temperature was lowered to –10 or –40 °C (entries 13 and 14)

Table 1. Condition Optimization of [2 + 2] Cycloaddition



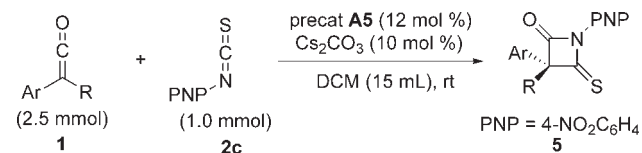
entry	precat.	DCM (x mL)	T (°C)	yield ^a (%)	ee ^b
1	A1	5	rt	69	39
2	A1	2.5	rt	60	33
3	A1	10	rt	70	52
4	A1	15	rt	96	55
5	A1	20	rt	91	57
6	A2	15	rt	74	55
7	A3	15	rt	75	85
8	A4	15	rt	66	83
9	A5	15	rt	73	96
10	B	15	rt	31	–33
11 ^c	A5	15	rt	76	96
12 ^{c,d}	A5	15	rt	80	97
13	A5	15	–10	60 (6) ^e	97 (96) ^e
14	A1	5	–40	<10 (18) ^e	42 (71) ^e

^a Isolated yield. ^b Determined by chiral HPLC. ^c Ketene **1a** was added via a syringe pump over 3 h. ^d 2.5 mmol of ketene **1a** was used. ^e Yield and ee of **7a** were showed in parentheses.

A variety of aryl(alkyl)ketenes were tested under the optimized conditions for the [2 + 2] cycloaddition (Table 2). Both aryl(alkyl)ketenes with electron-withdrawing groups (Ar = 4-Cl, 4-BrC₆H₄) and with electron-donating groups (Ar = 4-Me, 4-MeOC₆H₄) worked well to give the corresponding 4-thioxo-2-azetidiones **5** in good yields with high enantioselectivities (entries 2–5). The ketene **1f** with an *m*-chlorophenyl group worked well for the reaction, but

ketene **1g** with an *o*-chlorophenyl group did not (entries 6 and 7). The ketenes with methyl, *n*-propyl, and *n*-butyl groups all worked well, affording the desired cycloadducts in high yields with high enantioselectivities (entries 8–10). However, ketene **1k** with isopropyl group gave no cycloadduct under the current reaction conditions (entry 11).

Table 2. NHC-Catalyzed [2 + 2] Cycloaddition



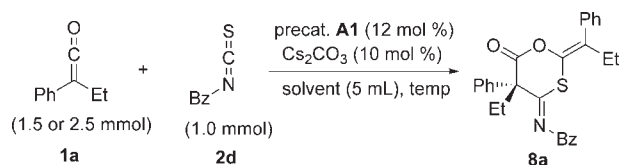
entry	1 (Ar, R)	5	time (h)	yield ^a (%)	ee ^b (%)
1	1a , Ph, Et	5a	21	80	97
2	1b , 4-ClC ₆ H ₄ , Et	5b	21	75	97
3	1c , 4-BrC ₆ H ₄ , Et	5c	27	62	95
4	1d , 4-MeC ₆ H ₄ , Et	5d	22	65	93
5	1e , 4-MeOC ₆ H ₄ , Et	5e	141	63	93
6	1f , 3-ClC ₆ H ₄ , Et	5f	39	70	96
7	1g , 2-ClC ₆ H ₄ , Et	5g	61	NR	/
8	1h , Ph, Me	5h	15	78	92
9	1i , Ph, <i>n</i> -Pr	5i	20	81	93
10	1j , Ph, <i>n</i> -Bu	5j	10	85	92
11	1k , 4-ClC ₆ H ₄ , <i>i</i> -Pr	5k	125	NR	/

^a Isolated yield. ^b Determined by HPLC. NR = no reaction.

We then turned our attention to the possible [2 + 2 + 2] cycloaddition (Table 3). It is interesting that although the reaction with *N*-benzoylisothiocyanate gave a complex mixture at room temperature, the corresponding [2 + 2 + 2] cycloadduct **8a** was isolated in 40% yield with 86% ee without the observation of [2 + 2] cycloadduct when the reaction was carried out at $-10\text{ }^{\circ}\text{C}$ (entries 1 and 2). Solvent screening revealed that no improvement was made in toluene or acetonitrile (entries 3 and 4). Reaction temperature (-10 to $-60\text{ }^{\circ}\text{C}$) and dosage of the ketene had strong impact on the yield of the reactions (entries 5–8). A little excess of ketene (**1a:2d** = 2.5:1) and lowering the reaction temperature to $-40\text{ }^{\circ}\text{C}$ gave cycloadduct **8a** in 60% yield with 83% ee (entry 7). Further experiments showed that concentration could also alter the yield of the reaction (entries 7, 9–10). Moderately concentrated solution (**[1a]** = 1.0 M) benefited the reaction, giving cycloadduct **8a** in 72% yield with 85% ee (entry 9).

Several ketenes were then tested for the [2 + 2 + 2] cycloaddition reaction with *N*-benzoyl isothiocyanate, which showed similar scope as the [2 + 2] cycloaddition reaction with *N*-*p*-nitrophenyl isothiocyanate (Table 4). Both electron-withdrawing substituents and electron-donating substituents were tolerable for the aryl(alkyl)-ketenes (entries 2–5). However, ketene **1g** with a 2-chlorophenyl group gave no cycloadduct (entry 6). The ketenes with methyl, *n*-propyl, and *n*-butyl groups all worked well, but ketene **1k** with isopropyl group did not (entries 7–10).

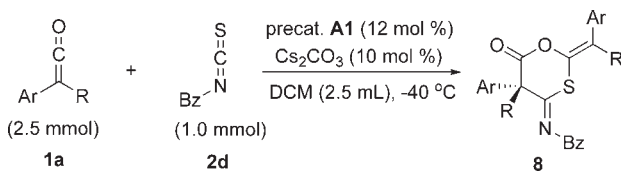
Table 3. Condition Optimization of the [2 + 2 + 2] Cycloaddition



entry	1a:2d	solvent	<i>T</i> (°C)	yield ^a (%)	ee ^b (%)
1	1.5:1	DCM	rt	mixture	/
2	1.5:1	DCM	-10	40	86
3	1.5:1	toluene	-10	21	87
4	1.5:1	CH ₃ CN	-10	15	46
5	1:1.5	DCM	-10	21	77
6	2.5:1	DCM	-10	40	73
7	2.5:1	DCM	-40	60	83
8	2.5:1	DCM	-60	39	84
9 ^c	2.5:1	DCM	-40	72	85
10 ^d	2.5:1	DCM	-40	64	83

^a Isolated yield. ^b Determined by HPLC. ^c 2.5 mL of DCM was used. ^d 1.5 mL of DCM was used.

Table 4. NHC-Catalyzed [2 + 2 + 2] Cycloaddition



entry	1 (Ar, R)	8	time (h)	yield ^a (%)	ee ^b (%)
1	1a , Ph, Et	8a	45	72	86
2	1b , 4-ClC ₆ H ₄ , Et	8b	53	62	75
3	1d , 4-MeC ₆ H ₄ , Et	8d	41	71	78
4	1e , 4-MeOC ₆ H ₄ , Et	8e	78	54	67
5 ^c	1f , 3-ClC ₆ H ₄ , Et	8f	39	50	61
6 ^d	1g , 2-ClC ₆ H ₄ , Et	8g	50	NR	/
7	1h , Ph, Me	8h	38	60	77
8	1i , Ph, <i>n</i> -Pr	8i	51	65	81
9	1j , Ph, <i>n</i> -Bu	8j	31	66	79
10 ^d	1k , 4-ClC ₆ H ₄ , <i>i</i> -Pr	8k	113	NR	/

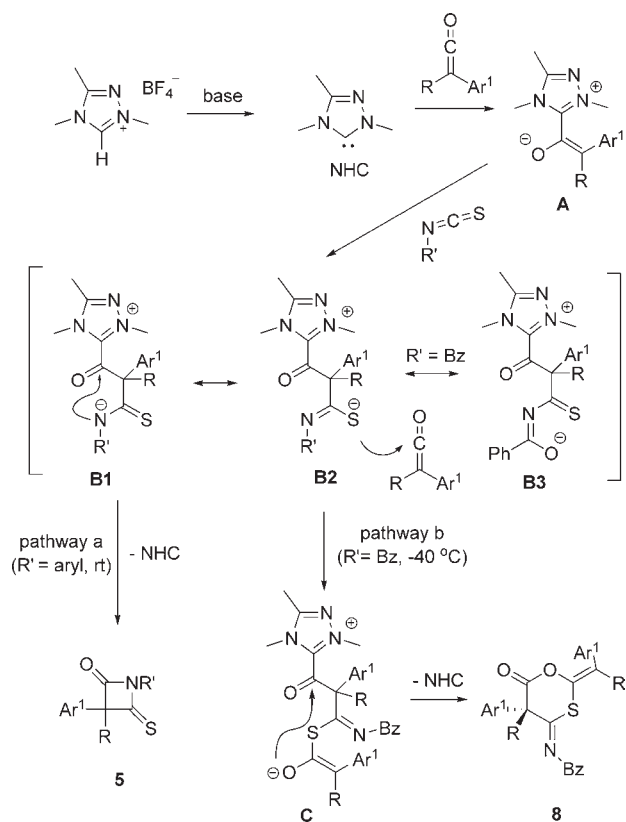
^a Isolated yield. ^b Determined by HPLC. ^c Reaction was carried out at $-10\text{ }^{\circ}\text{C}$. ^d No reaction occurred at $-40\text{ }^{\circ}\text{C}$ or room temperature.

Both the structures of [2 + 2] cycloadduct **5h** and [2 + 2 + 2] cycloadduct **7a** were unambiguously established by X-ray analysis of their crystals (Figures S1 and S3, Supporting Information).¹⁵

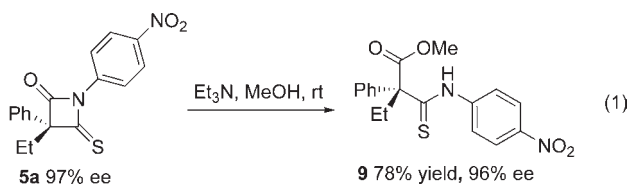
The resulted highly functionalized cycloadducts offer many opportunity of chemical transformations. For example, alcoholysis of the cycloadduct **5a** gave the ring-opening

(15) See the Supporting Information for details.

Scheme 2. Plausible Reaction Pathways



compound **9** in 78% yield without apparent erosion of enantiopurity (eq 1).



The plausible reaction pathways are depicted in Scheme 2. The addition of NHC to ketenes generates intermediate **A**,

which reacts with isothiocyanate to give intermediate **B**. Intermediate **B** may collapse via an intramolecular *N*-addition to azolium acyl to furnish [2 + 2] cycloadducts **5** and regenerates NHC (pathway a). However, when the isothiocyanate with *N*-benzoyl is employed the intermediate **B** is stabilized with resonant structure **B3**, which facilitates the *S*-addition to the second molecule of ketene to generate intermediate **C** at low temperature (pathway b). Intermediate **C** collapses via an intramolecular *O*-addition to azolium acyl to furnish [2 + 2 + 2] cycloadducts **8** and regenerates NHC. Control experiments revealed that [2 + 2] cycloadduct could not be transformed to [2 + 2 + 2] cycloadduct in the presence of ketene under the reaction condition of [2 + 2 + 2] cycloaddition.¹⁵

In summary, the enantioselective N-heterocyclic carbene-catalyzed formal [2 + 2] and [2 + 2 + 2] cycloadditions of ketenes and isothiocyanates were developed. Reaction with *N*-arylisothiocyanates at room temperature favors the [2 + 2] cycloaddition to give the 4-thioxo-2-azetidiones (thioxo- β -lactams) in good yields with high enantioselectivities, while reaction with *N*-benzoyl isothiocyanates at -40 °C favors the [2 + 2 + 2] cycloaddition to give the corresponding 1,3-oxathian-6-ones in good yields with good enantioselectivities. The resulted optically active cycloadducts are potentially useful for the synthesis of sulfur-containing heterocycles, particularly the β -lactam antibiotics.

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Supporting Information Available. Experimental procedures, compound characterization and X-ray data for cycloadducts **5h** and **7a** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.